

## **EQUIPMENT REQUIRED:**

pH Meter 250 ml Beaker Stir Plate and Stir Bar Digital Electronic Scale (accurate to 1 g) and Weigh Boat 25 ml Buret 5 ml Pipette

# **REAGENTS:**

.0667 N Sodium Hydroxide (NaOH) Distilled Water

# SAMPLE PREPARATION:

Wine and juice sample should be degassed prior to testing.

# **PROCEDURE:**

### Concentrate:

- 1. Fill buret with the .0667 N Sodium Hydroxide.
- 2. Place 100 ml distilled water and a stir bar into a 250 ml beaker. Place on stir plate under the buret. Place pH electrode into water and turn on the stir plate. Add 5 ml of concentrate to the beaker to buffer the water.
- 3. Titrate the solution with NaOH to a pH of 8.20.
- 4. Weigh out 5 g of sample into a weigh boat. Carefully wash this into the buffer water, rinse and mix well so that all of the concentrate is in solution.
- 5. Refill the buret and retitrate to a pH of 8.20. Read buret and use following formula:

<u>(V)(N)(75)(100)</u> (1000)(W)

V=Volume of NaOH • N=Normality of NaOH • W= Weight of sample in g

## NOTE:

Because the sample is in grams, the answer will be in grams per 100 grams of sample.

# Example:

Titration of 13 ml: 13 x .0667 x 75 x 100 = 1.3

### **PROCEDURE:**

### Juice and Wine:

- 1. Fill buret with the .0667 N Sodium Hydroxide.
- 2. Place 100 ml distilled water and a stir bar into a 250 ml beaker. Place on stir plate under the buret. Place pH electrode into water and turn on the stir plate. Add 5 ml of juice or wine to the beaker to buffer the water.
- 3. Titrate the solution with NaOH to a pH of 8.20.
- 4. Pipette 5 ml of sample into the beaker, allow to mix well.
- 5. Refill the buret and retitrate to a pH of 8.20. Read buret and use following formula:

### <u>(V)(N)(75)(100)</u> (1000)(W)

V=Volume of NaOH • N=Normality of NaOH • W= Weight of sample in g

## NOTE:

Because the sample is in milliliters, the answer will be in grams per 100 milliliters of sample.

## Example:

Titration of 9 ml:  $9 \times .0667 \times 75 \times 100 = .09$ 1000 x 5

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